metal-organic compounds

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Di-*µ*-hydroxo-bis(diisopropylgallium)– 1,4,8,11-tetramethyl-1,4,8,11-tetraazacyclotetradecane (1/1)

Neil M. Boag,^a* Kathleen M. Coward,^a Anthony C. Jones,^b Martyn E. Pemble^a and J. Robin Thompson^a

^aDepartment of Chemistry, University of Salford, Salford M5 4WT, England, and ^bDepartment of Chemistry, University of Liverpool, Liverpool L69 72D, England Correspondence e-mail: n.m.boag@salford.ac.uk

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The hydrolysis product $[Ga_2(C_3H_7)_4(OH)_2] \cdot C_{14}H_{32}N_4$, derived from the tetrakis(triisopropylgallium)–1,4,8,11-tetramethyl-1,4,8,11-tetraazacyclotetradecane (1/1) adduct, consists of a centrosymmetric $[^iPr_2Ga(\mu - OH)]_2$ unit hydrogen bonded through the hydroxyl group to a nitrogen on an adjacent centrosymmetric 1,4,8,11-tetramethyl-1,4,8,11-tetraazacyclotetradecane molecule, resulting in the generation of a molecular chain through the crystal.

Comment

The product of controlled hydrolysis of gallium trialkyls depends on the steric bulk of the alkyl group and may result in tetramers, $[Me_2Ga(\mu_2-OH)]_4$ (Smith & Hoard, 1959), trimers, $[{}^{t}Bu_{2}Ga(\mu_{2}\text{-}OH)]_{3}$ (Naiini *et al.*, 1993; Atwood *et al.*, 1993), or dimers, $[R_2Ga(\mu-OH)]_2$ $[R = CH(SiMe_3)_2$, mesityl, C₆H₄-2-CH₂NMe₂] (Uhl et al., 1996; Storre et al., 1996; Coggin et al., 1993). Increasing the steric congestion by substitution of the OH group by an alkoxide also results in a decrease in the degree of oligomerization, e.g. $[{}^{t}Bu_{2}Ga(\mu - O'Bu)]_{2}$ (Cleaver et al., 1994). This latter effect may apparently be mimicked by hydrogen bonding of the bridging hydroxyl species to an appropriate base, e.g. $[{Me_3Ga(\mu-OH)}_3\cdot 3H_2O]_2\cdot 18$ -crown-6 (Croucher et al., 1999). This feature has allowed us to determine the molecular structure of $[{}^{i}Pr_{2}Ga(\mu-OH)]_{2}$ as a hydrogen-bonded adduct, (I), of 1,4,8,11-tetramethyl-1,4,8,11tetraazacyclotetradecane.



The molecule of (I) (Fig. 1) contains a centrosymmetric $Ga_2(\mu$ -OH)_2 four-membered ring with gallium-bonded isopropyl groups above and below the plane of the ring. The hydroxyl H atoms are located 0.38 (2) Å above/below this plane, which subtends an angle of $\sim 152^{\circ}$ to the O–H vector.

The structure is analogous to other $[R_2Ga(\mu-OH)]_2$ species and also to the chloride-substituted compounds $[R(Cl)Ga(\mu-OH)]_2$ $[R = 2,4,6^{-i}Pr_3Ph, 2,4,6-(CF_3)_3Ph$ or $(Me_3Si)_3Si]$ (Twamley & Power, 1999; Schluter *et al.*, 1994; Linti *et al.*, 1996).

The intra-annular Ga-Ga vector, 2.9425 (7) Å, is the shortest of the reported dialkyl derivatives, $[R_2Ga(\mu-OH)]_2$ (2.970–3.064 Å), presumably due to reduced steric hindrance. Accordingly, the internal angle subtended at gallium, $81.04~(6)^{\circ}$, is larger than that found in the other six examples of this structural type (78.8-80.6°) and the corresponding angle at oxygen, 98.96 (6) $^{\circ}$, is the smallest (99.1–102.1 $^{\circ}$). The Ga-O bond lengths, 1.9339 (12) and 1.9368 (12) Å, are compatible with the range of Ga–O bonds found in the other four-membered-ring species, 1.893-1.969 Å. However, the Ga-O bonds in the chloride-substituted derivatives, $[R(Cl)Ga(\mu-OH)]_2$, are generally shorter, which leads to shorter intra-annular Ga-Ga vectors (2.887-2.939 Å). The Ga-C bonds are some 0.03 Å shorter than the average found for adducts of ⁱPr₃Ga (Coward et al., 2000), where the bond length sequence $Ga-Me < Ga-Et < Ga^{i}Pr$ was demonstrated. However, they are comparable to those in other μ_2 -OH gallium species, such as $[Me_2Ga(\mu_2-OH)]_4$ (Smith & Hoard, 1959) and ['Bu₂Ga(μ_2 -OH)]₃ (Atwood *et al.*, 1993), although there are insufficient accurately determined data to confirm a similar sequence.

The four N atoms of the tetraazacyclotetradecane ring are crystallographically constrained to be coplanar. This plane subtends an angle of 82.71 (5)° to the plane of the Ga₂O₂ ring. Two independent ring configurations were found in the crystal structure of free 1,4,8,11-tetramethyl-1,4,8-11-tetraazacyclotetradecane, namely *RSSR trans*-IV and *RRSS trans*-III (Willey *et al.*, 1993; Kelly *et al.*, 1996). The tetraazacyclotetradecane ring in the gallium adduct adopts the higher



Figure 1

The molecular structure of (I) with displacement ellipsoids drawn at the 50% probability level.

energy *RRSS trans*-III configuration, which facilitates strong hydrogen bonding between N2 and the hydroxyl hydrogen, 2.03 (2) Å. The hydrogen bonds are approximately parallel to the *c* axis, resulting in a molecular chain running throughout the crystal. Apart from a possible intra-annular interaction between N1 and H3A (2.40 Å), there are no non-classical hydrogen bonds present.

Experimental

The 4:1 adduct of triisopropylgallium and 1,4,8,11-tetramethyl-1,4,8,11-tetraazacyclotetradecane (Coward *et al.*, 2000) was dissolved in dry toluene and left to crystallize at 243 K. After one week, there was evident decomposition, presumably due to hydrolysis. A few well shaped crystals present in the mixture were isolated and dried in a stream of dinitrogen.

Z = 1

 $D_x = 1.209 \text{ Mg m}^{-3}$

Cell parameters from 32

Mo $K\alpha$ radiation

reflections

 $\theta=5.44{-}15.04^\circ$

 $\mu = 1.654 \text{ mm}^{-1}$

Block, colourless

 $0.45 \times 0.30 \times 0.25 \text{ mm}$

T = 223 (2) K

Crystal data

$$\begin{split} & [\text{Ga}_2(\text{C}_3\text{H}_7)_4(\text{OH})_2]\cdot\text{C}_{14}\text{H}_{32}\text{N}_4 \\ & M_r = 602.24 \\ & \text{Triclinic, } P\overline{1} \\ & a = 8.5877 \ (7) \text{ Å} \\ & b = 9.7161 \ (17) \text{ Å} \\ & c = 11.1425 \ (10) \text{ Å} \\ & \alpha = 113.138 \ (8)^{\circ} \\ & \beta = 97.516 \ (4)^{\circ} \\ & \gamma = 98.551 \ (7)^{\circ} \\ & V = 826.92 \ (18) \text{ Å}^3 \end{split}$$

Data collection

Siemens P4 diffractometer $R_{\rm int}=0.018$ Profile fitting of $\theta/2\theta$ scans $\theta_{\rm max} = 27.5^{\circ}$ Absorption correction: ψ scan $h = -1 \rightarrow 11$ $k = -11 \rightarrow 11$ (XPREP in SHELXTL-Plus; Siemens, 1995) $l = -14 \rightarrow 14$ $T_{\rm min}=0.568,\ T_{\rm max}=0.661$ 3 standard reflections 4385 measured reflections every 97 reflections 3645 independent reflections intensity decay: 5% 3277 reflections with $I > 2\sigma(I)$

Refinement

Refinement on F^2	H atoms treated by a mixture of
R(F) = 0.026	independent and constrained
$wR(F^2) = 0.065$	refinement
S = 1.036	$(\Delta/\sigma)_{\rm max} = 0.001$
3645 reflections	$\Delta \rho_{\rm max} = 0.34 \text{ e } \text{\AA}^{-3}$
164 parameters	$\Delta \rho_{\rm min} = -0.22 \text{ e} \text{ Å}^{-3}$

The H atoms were placed in calculated positions and allowed to ride except for the hydroxyl hydrogen which was identified from a difference map and allowed to refine freely.

Data collection: *XSCANS* (Siemens, 1996); cell refinement: *XSCANS*; data reduction: *XSCANS*; program(s) used to solve structure: *SHELXS*97 (Sheldrick, 1997); program(s) used to refine structure: *SHELXL*97 (Sheldrick, 1997); molecular graphics: *SHELXTL-Plus* (Siemens, 1995); software used to prepare material for publication: *SHELXL*97.

Table 1

Selected geometric parameters (Å, °).

$ \begin{array}{ccc} Ga-O^{i} & 1.9339 (12) \\ Ga-O & 1.9368 (12) \\ \end{array} $		Ga–C6 Ga–Ga ⁱ	1.9872 (18) 2.9425 (7)
Ga-C7	1.9853 (18)		
O ⁱ -Ga-O	81.04 (6)	O-Ga-C6	112.91 (7)
O ⁱ -Ga-C7	112.84 (7)	C7-Ga-C6	123.30 (8)
O-Ga-C7	109.58 (7)	Ga ⁱ –O–Ga	98.96 (6)
O ⁱ -Ga-C6	109.36 (7)		

Symmetry code: (i) 1 - x, 1 - y, 1 - z.

Table 2

		0	
Hydrogen-bonding	geometry	(A,	°).

$D - H \cdots A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - H \cdots A$
O−H1···N2	0.80 (2)	2.03 (2)	2.8119 (19)	167 (2)

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Supplementary data for this paper are available from the IUCr electronic archives (Reference: JZ1426). Services for accessing these data are described at the back of the journal.

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