metal-organic compounds

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Di- μ -hydroxo-bis(diisopropylgallium) – 1,4,8,11-tetramethyl-1,4,8,11-tetraazacyclotetradecane (1/1)

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The hydrolysis product $[Ga_2(C_3H_7)_4(OH)_2] \cdot C_{14}H_{32}N_4$, derived from the tetrakis(triisopropylgallium) $-1,4,8,11$ -tetramethyl-1,4,8,11-tetraazacyclotetradecane (1/1) adduct, consists of a centrosymmetric $[{}^{i}Pr_{2}Ga(\mu$ -OH)]₂ unit hydrogen bonded through the hydroxyl group to a nitrogen on an adjacent centrosymmetric 1,4,8,11-tetramethyl-1,4,8,11-tetraazacyclotetradecane molecule, resulting in the generation of a molecular chain through the crystal.

Comment

The product of controlled hydrolysis of gallium trialkyls depends on the steric bulk of the alkyl group and may result in tetramers, $[Me₂Ga(μ ₂-OH)]₄ (Smith & Hoard, 1959), trimers,$ $[$ 'Bu₂Ga(μ ₂-OH)]₃ (Naiini *et al.*, 1993; Atwood *et al.*, 1993), or dimers, $[R_2Ga(\mu\text{-}OH)]_2$ $[R = CH(SiMe_3)_2$, mesityl, C_6H_4 -2- $CH₂NMe₂$] (Uhl et al., 1996; Storre et al., 1996; Coggin et al., 1993). Increasing the steric congestion by substitution of the OH group by an alkoxide also results in a decrease in the degree of oligomerization, e.g. $[{}^t\text{Bu}_2\text{Ga}(\mu\text{-O}'\text{Bu})]_2$ (Cleaver et al., 1994). This latter effect may apparently be mimicked by hydrogen bonding of the bridging hydroxyl species to an appropriate base, e.g. $[\{Me₃Ga(\mu-OH)\}₃3H₂O]₂18-crown-6$ (Croucher et al., 1999). This feature has allowed us to determine the molecular structure of $[{}^{i}Pr_{2}Ga(\mu$ -OH)]₂ as a hydrogen-bonded adduct, (I), of 1,4,8,11-tetramethyl-1,4,8,11 tetraazacyclotetradecane.

The molecule of (I) (Fig. 1) contains a centrosymmetric $Ga_2(\mu\text{-}OH)$ ₂ four-membered ring with gallium-bonded isopropyl groups above and below the plane of the ring. The hydroxyl H atoms are located $0.38(2)$ Å above/below this plane, which subtends an angle of $\sim 152^\circ$ to the O–H vector.

The structure is analogous to other $[R_2Ga(\mu\text{-}OH)]_2$ species and also to the chloride-substituted compounds $[R(\text{Cl})Ga(\mu-$ OH)]₂ [R = 2,4,6-^{*i*}Pr₃Ph, 2,4,6-(CF₃)₃Ph or (Me₃Si)₃Si] (Twamley & Power, 1999; Schluter et al., 1994; Linti et al., 1996).

The intra-annular Ga $-Ga$ vector, 2.9425 (7) Å, is the shortest of the reported dialkyl derivatives, $[R_2Ga(\mu\text{-}OH)]_2$ $(2.970-3.064 \text{ Å})$, presumably due to reduced steric hindrance. Accordingly, the internal angle subtended at gallium, 81.04 $(6)^\circ$, is larger than that found in the other six examples of this structural type $(78.8-80.6^{\circ})$ and the corresponding angle at oxygen, 98.96 (6) $^{\circ}$, is the smallest (99.1–102.1 $^{\circ}$). The Ga–O bond lengths, 1.9339 (12) and 1.9368 (12) Å, are compatible with the range of $Ga-O$ bonds found in the other four-membered-ring species, $1.893-1.969$ Å. However, the $Ga-O$ bonds in the chloride-substituted derivatives, $[R(C)Ga(\mu\text{-}OH)]_2$, are generally shorter, which leads to shorter intra-annular $Ga-Ga$ vectors $(2.887-2.939 \text{ Å})$. The $Ga-C$ bonds are some 0.03 Å shorter than the average found for adducts of ^{*i*}Pr₃Ga (Coward *et al.*, 2000), where the bond length sequence $Ga-Me < Ga-Et < Ga-{}^{i}Pr$ was demonstrated. However, they are comparable to those in other μ_2 -OH gallium species, such as $[Me₂Ga(\mu₂-OH)]_4$ (Smith & Hoard, 1959) and $[$ 'Bu₂Ga(μ ₂-OH)]₃ (Atwood *et al.*, 1993), although there are insufficient accurately determined data to confirm a similar sequence.

The four N atoms of the tetraazacyclotetradecane ring are crystallographically constrained to be coplanar. This plane subtends an angle of 82.71 (5) \degree to the plane of the Ga₂O₂ ring. Two independent ring configurations were found in the crystal structure of free 1,4,8,11-tetramethyl-1,4,8-11-tetraazacyclotetradecane, namely RSSR trans-IV and RRSS trans-III (Willey et al., 1993; Kelly et al., 1996). The tetraazacyclotetradecane ring in the gallium adduct adopts the higher

Figure 1 The molecular structure of (I) with displacement ellipsoids drawn at the 50% probability level.

energy RRSS trans-III configuration, which facilitates strong hydrogen bonding between N2 and the hydroxyl hydrogen, 2.03 (2) Å. The hydrogen bonds are approximately parallel to the c axis, resulting in a molecular chain running throughout the crystal. Apart from a possible intra-annular interaction between N1 and H3A (2.40 Å), there are no non-classical hydrogen bonds present.

Experimental

The 4:1 adduct of triisopropylgallium and 1,4,8,11-tetramethyl-1,4,8,11-tetraazacyclotetradecane (Coward et al., 2000) was dissolved in dry toluene and left to crystallize at 243 K. After one week, there was evident decomposition, presumably due to hydrolysis. A few well shaped crystals present in the mixture were isolated and dried in a stream of dinitrogen.

 $Z = 1$

 $D_r = 1.209$ Mg m⁻³

Cell parameters from 32

Mo $K\alpha$ radiation

reflections

 $\theta = 5.44 - 15.04^{\circ}$

 $\mu = 1.654$ mm⁻¹

Block, colourless

 $0.45\,\times\,0.30\,\times\,0.25$ mm

 $T = 223$ (2) K

Crystal data

 $[Ga_2(C_3H_7)_4(OH)_2]\cdot C_{14}H_{32}N_4$ $M_{\rm r} = 602.24$ Triclinic, $P\overline{1}$ $a = 8.5877(7)$ Å $b = 9.7161(17)$ Å $c = 11.1425(10)$ Å $\alpha = 113.138(8)^{\circ}$ $\beta = 97.516(4)$ ° $\gamma = 98.551(7)$ ° $V = 826.92(18)$ Å³

Data collection

Siemens P4 diffractometer $R_{\rm int} = 0.018$ Profile fitting of $\theta/2\theta$ scans $\theta_{\text{max}} = 27.5^{\circ}$ $h = -1 \rightarrow 11$ Absorption correction: ψ scan $k=-11\rightarrow11$ (XPREP in SHELXTL-Plus: Siemens, 1995) $l = -14 \rightarrow 14$ $T_{\min} = 0.568, T_{\max} = 0.661$ 3 standard reflections 4385 measured reflections every 97 reflections 3645 independent reflections intensity decay: 5% 3277 reflections with $I > 2\sigma(I)$

Refinement

The H atoms were placed in calculated positions and allowed to ride except for the hydroxyl hydrogen which was identified from a difference map and allowed to refine freely.

Data collection: XSCANS (Siemens, 1996); cell refinement: XSCANS: data reduction: XSCANS: program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: *SHELXTL-Plus* (Siemens, 1995); software used to prepare material for publication: SHELXL97.

Table 1

Selected geometric parameters (\mathring{A}, \degree) .

Symmetry code: (i) $1 - x$, $1 - y$, $1 - z$.

Table 2

Hydrogen-bonding geometry (\AA, \degree) .

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Supplementary data for this paper are available from the IUCr electronic archives (Reference: JZ1426). Services for accessing these data are described at the back of the journal.

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